Multivalent Molecules WS Name _____

Fill in the missing spaces in the charts below.

Chemical Formula	Chemical Name (Roman numeral)
HgO	**Murcury (I) oxide**
Cu_4C	Copper(I) Carbide
Mn_3P_2	**Manganese (II) Phosphide**
FeS	Iron (II) Sulfide
NiO	**Nickel (II) Oxide**
AsF_5	Arsenic (V) Fluoride
PbC	**Lead (IV) Carbide**
$SnCl_2$	Tin (II) Chloride
HgBr	**Mercury (I) Bromide**
NiP	Nickel (III) Phosphide
PbS	**Lead (II) Sulfide**
MnI_4	Manganese (IV) Iodide
Mn_3As_7	**Manganese (VII) Arsenide**
$AsBr_3$	Arsenic (III) Bromide
Cu_3N_2	**Copper (II) Nitride**
CoP	Cobalt (III) Phosphide
FeH_3	**Iron (III) Hydride**
SbN	Antimony (III) Nitride
$0Co_2C$	**Cobalt (II) Carbide**
MnF_4	Manganese (IV) Fluoride

Chemical Formula	Chemical Name (-ous/-ic method)
FeS	Ferrous Sulphide
HgO	**Murcuric oxide**
Cu_4C	Copperous Carbide
SnI_4	**Stannic Iodide**
FeS	Ferric Sulfide
NiO	**Nickelous Oxide**
AsF_3	Arsenous Fluoride
PbC	**Plumbic Carbide**
$SnCl_2$	Stannous Chloride
HgBr	**Murcurous Bromide**
NiP	Nickelic Phosphide
PbS	**plumbous Sulfide**
Hg_2O	Murcurous Oxide
$SnCl_2$	**Stannous Chloride**
$AsBr_3$	Arsenic Bromide
Cu_3N_2	**Cupric Nitride**
CoP	Cobaltic Phosphide
FeH_3	**Ferric Hydride**
SbN	Antimous Nitride
Co_2C	**Cobaltous Carbide**