

Chemistry Worksheet: Matter & Changes

1. A mixture (is/is not) a chemical combining of substances.
2. In a compound the (atoms/molecules) are (chemically/physically) combined so that the elements that make up the compound (retain/lose) their identities and (do/do not) take on a new set of properties.
3. The smallest identifiable unit of a compound is a(n) molecule, which is made up of elements which are chemically bonded.
4. True or False: A mixture is always made up of a combination of elements.
5. In a mixture, the substances (lose/retain) their identities.
6. In a mixture the substances involved (can/cannot) be separated by a simple physical process. In a compound the elements involved (can/cannot) be separated by a simple physical process because the elements are (physically combined/chemically bonded).
7. True or False: An element can be broken down into a simpler substance.
8. The smallest identifiable unit of an element is a(n) atom.
9. From the following list of substances, circle the ones that are elements: silver carbon dioxide wood, alcohol, chromium water, hydrogen carbon nitrogen oxygen gold sugar, salt, air, sulfur magnesium, nickel

INSTRUCTIONS: Classify each of the following changes in matter as physical [P] or chemical [C].

10. Grinding chalk into powder P

11. Burning gasoline C

12. Dissolving salt in water P

13. Hammering gold into foil P

14. Dissolving zinc in acid C

15. Melting ice P

16. Tearing a piece of paper P

17. Digesting food C

18. Stretching copper into wire P

19. Making hydrogen from water C

20. Breaking glass P

Elements, Compounds, and Mixtures

Classify each of the pictures below by placing the correct label in the blanks below:

A= Element

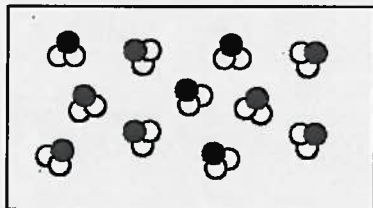
B= Compound

C= Mixture of elements

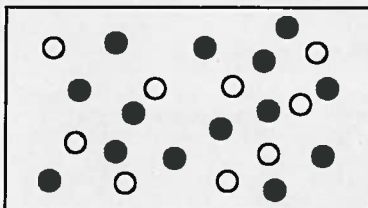
D= Mixture of compounds

E= Mixture of elements and compounds

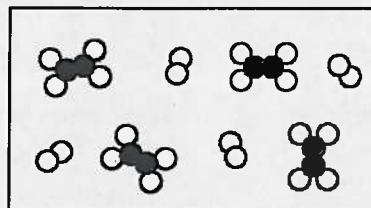
Each circle represents an atom and each different color represents a different kind of atom. If two atoms are touching then they are bonded together.



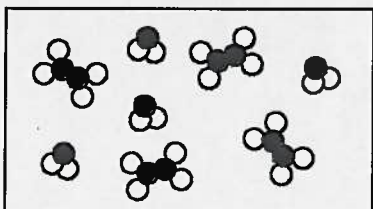
1) B



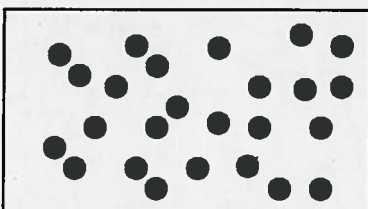
2) C



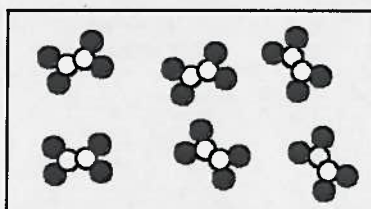
3) D



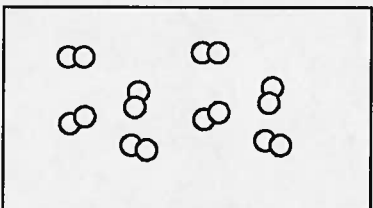
4) D



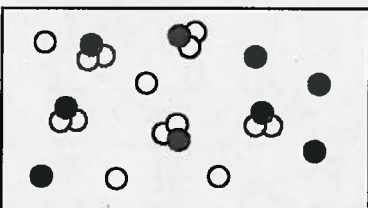
5) A



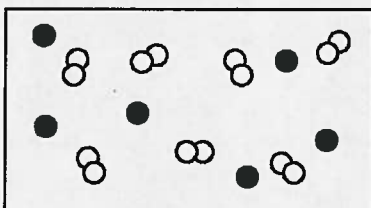
6) B



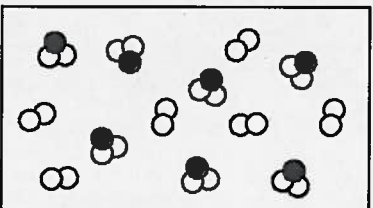
7) B



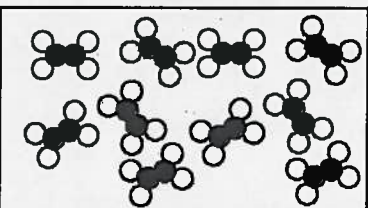
8) E



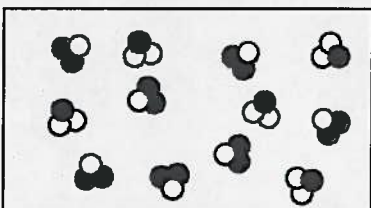
9) E



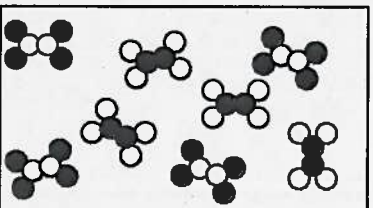
10) D



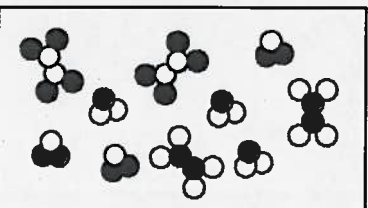
11) B



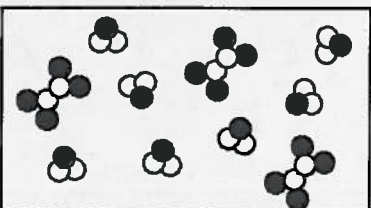
12) D



13) D



14) D



15) D

Physical and Chemical Changes

Name: ANSWERS

Date: _____ Hour: _____

Place a check in the appropriate column:

Change	Physical Change	Chemical Change
Salt dissolves in water.	✓	
Hydrochloric acid reacts with magnesium to produce hydrogen gas.		✓
A piece of copper is cut in half.	✓	
A sugar cube is ground up.	✓	
Water is heated and changed to steam.	✓	
Iron rusts.		✓
Ethyl alcohol evaporates.	✓	
Ice melts.	✓	
Milk sours (goes bad).		✓
Sugar dissolves in water.	✓	
Sodium and potassium react violently with water.		✓
Pancakes cook on a griddle.		✓
Grass grows on a lawn.		✓
A tire is inflated with air.	✓	
Food is digested in the stomach.		✓
Water is absorbed by a paper towel.	✓	
Ethyl alcohol boils at 79°C.	✓	
Paper burns.		✓
Water freezes at 0°C.	✓	
Fireworks explode.		✓
Alka-Seltzer gives off carbon dioxide when added to water.		✓
Clouds form in the sky.	✓	